An Introduction to Hydrogen From Production to End Uses

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Topics for today

- Why hydrogen?
- The basics
- Environmental, health, and safety
- Current markets
- Production methods
- Moving & storing
- Using hydrogen
- Economics overview



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Why hydrogen? The importance of hydrogen to a low-carbon future.

Decarbonization Pathways Enabled by Innovation

~15-30 years

Decarbonization

Accelerate economy-wide, low-carbon solutions

- Electric sector decarbonization
- Transmission and grid flexibility: storage, demand, EVs
- Efficient electrification

Achieve a net-zero clean energy system

- Ubiquitous clean electricity: renewables, advanced nuclear, CCUS
- Negative-emission technologies
- Low-carbon resources: hydrogen and related, low-carbon fuels, biofuels, and biogas



Reducing Economy-Wide CO₂ Emissions







Reducing Economy-Wide CO₂ Emissions





Today's Energy System





A Glimpse into the Future

New Resources and Players how will they fit and transition?



There are various types of climate risk

Physical risk: risk associated with changes in physical environmental and natural resource conditions with climate change Generally greater risk with greater change

Transition risk: risk associated with uncertain transitions to potential low-carbon futures May include impacts of regulatory, technological, or market changes to address climate change mitigation or adaptation

Society level: address transition risk now via mitigation efforts, reduce physical risk down the road



Informing company strategies to support risk mitigation through the energy transition

Company level: must contend with both—climate is changing, but so is the business and regulatory landscape

Different kinds of uncertainty for each

And...there is a cost



United States Billion-Dollar Disaster Events 1980-2021 (CPI-Adjusted)

EPRI 3002025872

Source: NOAA National Centers for Environmental Information (NCEI) U.S. Billion-Dollar Weather and Climate Distasters (2022).

The basics of hydrogen and its key properties

Basic Chemistry



Molecular Hydrogen: *H*₂

- Lightest element (H) on the periodic table
 - Contains one proton and one neutron
- Exists at a relatively low concentration in the atmosphere
 - Combines with other elements to form compounds:



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Unique Properties of Hydrogen

- Low volumetric energy density
 - Very high energy-to-weight ratio
 - More storage space needed. Compressed or liquified storage increases cost
- Can be liquified (LH₂)
 - Allows for much higher energy density compared to gaseous H₂
 - Energy-intensive process and LH₂ can be lost through evaporation
- Small molecular size
 - Causes H₂ to disperse a lot more quickly than other fuel
 - Need proper materials and tools to contain and detect
 - Leaks through joints & seals in pipes more easily than natural gas
- Hydrogen absorption by materials
 - Allows for storage of hydrogen using metal hydrides
 - Can embrittle steel, cause fatigue cracks, and degrade plastics
- Non-toxic
 - Exposure to other fuels & vapors can cause adverse health outcomes
 - Combustion of pure H₂ does not produce poisonous CO gas
 - Combustion still produces NO_x since N₂ and O₂ are the dominant constituents in air and react spontaneously at high temperatures



Photo by CEphoto, Uwe Aranas or alternatively © CEphoto, Uwe Aranas <u>CC BY-SA 4.0</u>



US Department of Energy (DOE). Available online: <u>https://www.energy.gov/eere/fuelcells/hydrogen-</u> storage (accessed on 02/25/23).



Environmental, health, and safety considerations of hydrogen





Environmental health and safety considerations

Hazards

Very small molecule: need proper materials to contain and detect

Material compatibility: can embrittle some metals, causing cracks and leaks

Fire: highly flammable, potential flame jetting. Pale blue flame when burning is difficult to see in daylight Explosive: high explosive energy

Pollutants: H_2 generates incremental NO_x when it is combusted alone or blended with natural gas

Detection: H₂ gas is colorless and odorless. Burns with pale blue flame which is difficult to see in daylight.
 Storage: Typically transported and stored at high pressure (gas) and very low temperature (liquid)

Asphyxiation: can occur in enclosed areas (unlikely)







Hydrogen and Propane Flames in Daylight (Photo courtesy of HAMMER) Hydrogen and Propane Flames at Night (Photo courtesy of ImageWorks)

Source: h2tools https://h2tools.org/bestpractices/hydrogen-flames

Proper design and procedures must be used to avoid potential fire or explosion

Environmental health and safety mitigations

Hazards	Mitigations	
Very small molecule: need proper materials to contain and detect	Welded connections and hydrogen-compatible materials. Consult industry standard for hydrogen-specific component requirements	
Material compatibility: can embrittle some metals, causing cracks and leaks	Codes & standards (ASME B31.12) outline material requirements for leak protection as well as to minimize embrittlement and corrosion	
Fire: highly flammable, potential flame jetting. Pale blue flame when burning is difficult to see in daylight Explosive: high explosive energy	Utilize leak and flame detectors , proper design and system shutdown planning per NFPA 2, industry best practices, control ignition sources, and regular maintenance	
Pollutants: H ₂ generates incremental NO _x when it is combusted alone or blended with natural gas	NOx emissions can be reduced by employing new burner designs & selective catalytic reduction systems for post-combustion control	
Detection: H ₂ gas is colorless and odorless. Burns with pale blue flame which is difficult to see in daylight.	H ₂ gas and flame detectors commercially available, with newer leak detection methods being explored (e.g., wide-area leak detection)	
Storage: Typically transported and stored at high pressure (gas) and very low temperature (liquid)	Gas storage in cylinders per DOT/FMVSS requirements (vehicles), or gas/liquid per ASME (stationary tank storage), with general designs per NFPA 2. Salt cavern storage being developed	
Asphyxiation: can occur in enclosed areas (unlikely)	Hydrogen sensors should be placed in enclosed spaces where hydrogen could collect	

Follow existing codes and standards and use best practices to minimize hydrogen risks

Analysis of H₂ Incidents in Industry



- Many incidents involve several contributing factors
 - Human errors are likely to cause an incident
 - Most occur after ignoring near misses & warnings
- A strong safety culture
 - must be established by leadership
 - Leads employees to work effectively and feel comfortable raising concerns

Approaches to Mitigate H₂ Incidents & Consequences

- Leak reduction:
 - Early detection via sensors
 - Automatic interlock & alarm activation mechanisms
 - Regular inspections and maintenance

Ignition minimization:

- Purging: use inert gas for testing equipment
- Areas electrical classification
- Ventilation design and management
- Equipment siting
- **Consequence reduction** (fire or explosion):
 - Maximize distance and shielding
- Safer process equipment:
 - Inherent safe design
 - Provide regular and up-to-date training

Many codes and standards are derived from proven best practices

Key Siting Considerations



- Electrical Source
 - Baseload operation versus renewable or grid support
 - Long term price contracts
- Water Source
 - Sustainable water sources
 - Long term water contracts
 - Equipment to process water
- Land Use and Footprint
 - Space available for system
- Transport or Conversion
 - Pipeline access, highway, shipping, or transport routes
 - Above or underground storage (salt caverns, etc.)
 - Footprint for chemical conversion (NH₃, etc.)
- Offtake and End Use
 - Offtake or end use customer
 - Long term contract potential
 - Price appetite considering production, storage, and delivery

Equity and Environmental Justice

EPA's definition

 Environmental justice is the fair treatment and meaningful involvement of all people, regardless of race, color, national origin, or income, with respect to the development, implementation, and enforcement of environmental laws, regulations, and policies

Goals

- An energy future that fully and meaningfully involves communities
- Social justice: An equitable distribution of both benefits and costs
- It is a part of decision making throughout the process
- Movement towards a just transition and a just future
- With expertise comes responsibility present accessible, understandable, and factual information

Helpful resources

- EPRI's Equitable Decarbonization Interest Group (EDIG)
- DOE's energy justice mapping tool: <u>https://energyjustice.egs.anl.gov/</u>
- DOE's disadvantaged communities mapping tool: <u>https://screeningtool.geoplatform.gov/en/#3.74/25.83/-93.2</u>
- DOE's Justice 40 initiative: <u>https://www.energy.gov/diversity/justice40-initiative</u>



Report #: 3002023584 Just Transition: An Overview of the Landscape and Leading Practices <u>https://www.epri.com/research/products/</u> 00000003002023584

Community Concerns of Hydrogen



The various hydrogen production techniques



Hydrogen Production Methods



Electrolysis using electricity from renewable energy sources



Steam methane reforming of natural gas



Steam methane reforming, but much of the emissions are captured and stored

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Production – Fossil Derived Hydrogen

Fossil-derived sources without carbon capture are the most prominent sources of hydrogen

- ~62% of production is from SMR (worldwide)
- ~19% of production is from coal gasification (worldwide)

Currently, SMR and gasification accounts for ~900 MMT of CO_2 /year

- SMR: emits ~10 metric tons of CO₂ for every metric ton of hydrogen produced
- Coal gasification: almost twice as CO₂intensive as SMR



'Green' Hydrogen Simplified



Hydrogen could decarbonize many end uses in addition to power generation technologies

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Production – Electrolysis

- Electrochemical process that splits water into hydrogen and oxygen using an electrolyzer powered by electricity
- Electrolyzers have a positively charged anode electrode generating oxygen and a negatively charge cathode electrode generating hydrogen, which are separated by a membrane or separator.



The methods for hydrogen storage and delivery





Transport, Storage, and Distribution Overview

- Transport methods
 - Pipelines
 - On-road vehicles
 - Shipping
- Storage methods
 - Underground (salt caverns, depleted oil and gas reservoirs, saline aquifers)
 - Tanks (compressed gas, liquified)
 - Converted to other energy carriers
- Considerations
 - Conversion costs
 - Compression vs. liquefaction
 - Transmission vs. distribution pipelines
 - Electricity costs

Source: DOE https://www.energy.gov/eere/fuelcells/hydrogen-pipelines





Fuel Ce





Source: NREL

Cavern

Moving & Storing = Energy



to 250 bar, higher pressures require special exemptions

Data source: <u>Elberry et al.</u> References: US DOE, EPRI



US natural gas pipeline and underground storage sites

Bulk Storage

- Geologic storage of H₂ can be used to:
 - Meet seasonal energy demand needs
 - Ensure continuity in supply during disruptions
 - Arbitrage low-cost energy to high demand times
 - Helps integration of more intermittent renewables
- A need for large scale, long duration storage
 - The U.S. natural gas infrastructure has immense energy storage capacity
 - Intense demand needs of seasonal space heating
 - Can meet ~16% of total annual natural gas demand
 - U.S. Electricity storage meets ~0.7% of typical annual electricity demand
 - >90% of this is from pumped hydro
 - Battery storage capacity is growing, yet small and lacks seasonal capabilities





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U.S. existing natural gas & H₂ potential storage capacity, TWh



Where is hydrogen used today?

Current Production Scale and Use

Worldwide Production ~94 million metric tons (MMT)



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United States Hydrogen Annual Usage ~10 million metric tons (MMT)



U.S. Department of Energy (DOE). "DOE National Clean Hydrogen Strategy and Roadmap," 2023. https://www.hydrogen.energy.gov/docs/hydrogenprogramlibraries/pdfs/us-national-clean-hydrogen-strategy-roadmap.pdf



A Closer Look at U.S. Hydrogen Supply and Demand

US State Total Daily Demand of Hydrogen Current Hydrogen Demands



Hydrogen Demand (kg/d) 8,000,000 6,000,000 4,000,000 2,000,000 U.S. State Average Delivered Price of Hydrogen, \$/kg Current Average Delivered Price of Hydrogen to End Users



Hydrogen Price (\$/kg)

15.0 12.5 10.0 7.5 5.0 2.5

Hydrogen demand varies geographically. Most hydrogen demand is concentrated in Texas, Louisiana, and California due to high concentrations of petroleum refining facilities. The areas with higher hydrogen demand are typically the areas where hydrogen prices are lower. This is primarily driven by proximity to hydrogen production facilities and demand volumes.



Hydrogen's applications across multiple end use sectors





Use Cases of Hydrogen to Decarbonize Final Energy





Opportunities for low-carbon fuels in the power sector

How can low-carbon fuels support peaking & reliability needs?



How can low-carbon fuels support the overall energy system?



Emerging markets

Potential export opportunities



Low-carbon fuels: leveraging new and existing markets





International & domestic opportunities



Emerging Domestic Energy Economy

Emerging International Energy Trade Flows



Source: IRENA 2022, "Global hydrogen trade to meet the 1.5°C climate goal: Trade outlook for 2050 and way forward". https://www.irena.org/publications/2022/Jul/Global-Hydrogen-Trade-Outlook



Hydrogen is key to the power-to-fuels value chain



Economic Considerations

Current U.S. Hydrogen Supply and Demand

U.S. Hydrogen Prices and Merchant Demand by Market Segments

Delivered price of hydrogen varies widely between and within industries. Processes with larger demands such as refining and ammonia production are able to realize lower costs due to higher volumes and lower purity requirements.

Smaller market segments with higher purity requirements on average, are subject to higher delivered prices of hydrogen. Depending on the specific volume and purity requirement this could range from \$2 to \$12/kg of hydrogen.



Demand volume weighted price averages by industry are shown on the chart above. Refining, chemical, and methanol industries currently have an average delivered price of hydrogen less than \$2/kg. While smaller hydrogen demand industries on average pay over \$4/kg.

New Market Development

Hydrogen Offtake Applications: What are you willing to pay?



U.S. Department of Energy (DOE). "DOE National Clean Hydrogen Strategy and Roadmap," 2023. https://www.hydrogen.energy.gov/docs/hydrogenprogramlibraries/pdfs/us-national-clean-hydrogen-strategy-roadmap.pdf

New Market Development

Clean Hydrogen Demand in Key Sectors: How much do you want?



U.S. Department of Energy (DOE). "DOE National Clean Hydrogen Strategy and Roadmap," 2023.

https://www.hydrogen.energy.gov/docs/hydrogenprogramlibraries/pdfs/us-national-clean-hydrogen-strategy-roadmap.pdf

New Market Development

When could adoption occur?



Clean hydrogen will meet demands in waves based on attractiveness in each end-use application

U.S. Department of Energy (DOE). "DOE National Clean Hydrogen Strategy and Roadmap," 2023. https://www.hydrogen.energy.gov/docs/hydrogenprogramlibraries/pdfs/us-national-clean-hydrogen-strategy-roadmap.pdf

Change is never easy Market headwinds & growing pains



Electrolyzer manufacturers struggle to turn growing order books into profits

Wednesday, September 6, 2023 4:04 AM ET

By Camilla Naschert

Market Intelligence



Source: IEA Hydrogen Projects Database. Paris, France: 2022.



Large potential = large impacts



Diverse range of end-use applications with large markets

Reference

Net-Zero

2050

2040

47

0

2020

6

4

2

GtCO₂

Pathways to Net-Zero LCRI U.S. NET-ZERO 2050

Reference with no new carbon policy,

continued technology improvements

2030

Net-Zero by 2050 with three core sensitivities around CCS, gas, bioenergy

	All Options	Higher Fuel Cost	Limited Options
Geologic Storage of CO ₂	Lower Costs	Higher Costs	Not Available
Natural Gas Supply Costs	Lower Costs	Higher Costs	Lower Costs
Bioenergy Feedstock Supply	Full	Supply Limited	Supply Limited

Source: LCRI Report <u>3002024882</u>



Full report available at lowcarbonlcri.com/netzero

Primary and Final Energy in Net-Zero 2050 Scenarios



Hydrogen Supply and Demand



Electrolytic Hydrogen Cost Sensitivities

LCOH = Levelized Cost of Hydrogen



influence on electrolytic hydrogen production costs



Electrolysis LCOH Comparisons: 45V Impacts (\$3/kg)



Potential to have significant impacts on H_2 production costs from electrolysis

2% O&M and PTC escalation, 5.4% discount rate, 20-year MACRS, 20-year booklife, 2% degradation

Challenges, benefits, & opportunities



Benefits of Low-Carbon Hydrogen

Reduced emissions and environmental impact

- No direct CO₂ emissions when combusted
- Decarbonize current sectors that use hydrogen
- Decarbonize hard-to-abate sectors that can be difficult to electrify
- Possibility to positively contribute to equity and environmental justice

Balancing asset

- Increased storage opportunities
- Able to capture peaks in electricity generation from renewables and store as hydrogen molecules for later
- Deliver hydrogen and other hydrogen carriers to areas that need the energy
- Increases reliability and resilience

Economic

- Spur domestic manufacturing
- Enable energy affordability
- Increase jobs
- Learning by doing and economies of scale
 - Scaling renewable energy can further decrease the cost of renewable energy as well as low-carbon hydrogen

Hydrogen can help make the case for integrating more renewable energy into the grid

U.S. Department of Energy (DOE). "DOE National Clean Hydrogen Strategy and Roadmap," 2023. https://www.hydrogen.energy.gov/docs/hydrogenprogramlibraries/pdfs/us-national-clean-hydrogen-strategy-roadmap.pdf

Creating a Market for Low-Carbon Hydrogen

Government Support through the Infrastructure Bill



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Public resources

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